

Ion implantation of Gd⁺, Er⁺ and Au⁺ into various crystallographic cuts of ZnO – structural and photoluminescence properties



J. Cajzl^{*1}, P. Nekvindová¹, K. Jeníčková¹, A. Jagerová^{2,3}, A. Macková^{2,3}, J. Oswald⁴, U. Kentsch⁵

¹ Department of Inorganic Chemistry, University of Chemistry and Technology, Prague, Czech Republic ² Nuclear Physics Institute, Czech Academy of Sciences, v. v. i., Řež, Czech Republic ³ Department of Physics, J.E. Purkinje University, Ústí nad Labem, Czech Republic ⁴ Institute of Physics, Czech Academy of Sciences, v.v.i., Prague, Czech Republic ⁵ Helmholtz-Zentrum Dresden-Rossendorf, Institute of Ion Beam Physics and Materials Research, Dresden, Germany

Zinc oxide (ZnO) is a wide-bandgap semiconductor ($E_g \approx 3.3 \text{ eV}$) hexagonal crystal with low phonon energy and very good physical and chemical stability [1]. Nowadays, ZnO crystal material is broadly studied in photonics for many applications. Doping of ZnO with various ions follows mainly two different motivations: enabling either new optical properties or new magnetic properties of the modified ZnO crystal. In this contribution, we report on the results of ion implantation of Gd⁺, Er⁺ and Au⁺ ions accelerated to 400 keV into the single-crystal ZnO wafers with <0001> crystallographic orientation (the implanted samples were subsequently annealed in oxygen). Ions used in this study were chosen for their different chemical nature as well as specific use in science and industry – erbium belong to the rare earth elements used for photonics and optical telecommunications, gadolinium is used because of its magnetic properties and gold is used in bio-applications. We focused particularly on the relationship between the two main intrinsic luminescence bands of ZnO – NBE peak and DLE band at around 375 and 530 nm, respectively. We studied in detail the possibility of controlling the luminescence properties of ZnO by varying the position of the short-wavelength bands.

EXPERIMENTS

Substrates

 Single crystalline zinc oxide (ZnO) wafers with <0001> crystallographic orientation

Ion implantation

- Implanted ions Gd⁺, Er⁺, Au⁺
- Beam energy 400 keV
- Fluences 5.0 x **10**¹⁴ -

5.0 x **10¹⁵ ions/cm²**

MEASUREMENTS

- Rutherford Backscattering Spectrometry (RBS) Er concentration depth profiles, Er positions, degree of damage
- Photoluminescence (PL) spectroscopy VIS/NIR intensity of luminescence in the visible and near-infrared spectral regions

Theoretical DFT calculations

• Used programs: CASTEP [2] and Accelrys Materials Studio

Er:ZnO



Au:ZnO

Theoretical results - DFT calculations (geometry optimization of local structure around Er, Au, Gd)



- Erbium: geometry optimizations showed that Er will be most probably located in substitutional position of Zn (or alternatively V_{oct.}).
- Gadolinium: theoretical calculations showed that Gd will be most probably located in octahedral void (or alternatively Zn_{subst.}).
- Gold: theoretical calculations showed that Au will be located in all four positions with the same (low) probability, i.e. Au is practically inert.

Experimental results – Fine scans of RBS Channeling (positions of Er, Au and Gd ions in the ZnO structure)



• Erbium: after annealing for lower fluences Er stayed in the Zn_{subst}

while for higher fluences Er shifted to interstitial positions.



 Gadolinium followed similar trend to erbium, i.e. after annealing for lower fluences Gd stayed while for higher shifted to interstitial positions.



 Gold ions were even for the lowest fluence before and after annealing located in interstitial positions. Moreover, gold in our samples formed Au nanoparticles (see HRTEM images).

Experimental results – Photoluminescence spectra (luminescence of Er-, Au- and Gd-doped ZnO samples)



350 375 400	450 500 550 Wavelength [nm]	600 350 400 450 500 550 600 650 Wavelength [nm]	350 400 450 500 550 600 650 Wavelength [nm]	VB
UV-VIS spectra showed that ion implantation causes vanishing of the excitonic NBE (near-band-edge) emission while subsequent annealing in oxygen causes recovery of the ZnO structure and therefore changes of the number of oxygen vacancies which showed in the spectra as				
		an increase of the excitonic NBE band and vanish of the DI	-E (deep-level-emission) band (connected with the vac	ancies in ZnO structure).

CONCLUSIONS

- According to the theoretical simulations and experimental angular scans, erbium and gadolinium ions in the single-crystal ZnO will most probably be located in the Zn substitutional positions and large octahedral voids. Gold ions showed their inert behaviour no preferential position from simulations and measurements.
- Annealing caused movement of all three ions from substitutional to interstitial positions which was more prominent for higher implantation fluences.
- The structural damage of ZnO was increasing with increasing implantation fluence in the range of approx. 10–30%.
- Due to inert noble-metal behaviour of Au (i.e. no willingness to form bonds with oxygen atom in ZnO), the Au nanoparticles were detected by HRTEM measurements.
- Ion implantation caused vanishing of luminescence of ZnO (mainly excitonic NBE peak), while after annealing at 600 °C in oxygen the excitonic peak of ZnO recovered. Samples with Au ions differed from Er and Gd samples in that Au samples had emission band at around 430 nm which can possibly be assigned to formation of vacancies and interstitials (Zn and Au).

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Corresponding author: cajzlj@vscht.cz

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